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Optoelectronic properties of SnO₂ / p-Si heterojunction prepared by a simple chemical bath deposition method

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Abstract: The SnO₂ film was successfully deposited on Si wafer by a simple chemical bath method to fabricate n-SnO₂/p-Si heterojunction structure photoelectric device. The self-made chemical bath system is very cheap and convenient. The structural, optical and electrical properies of the SnO₂ film were studied by XRD, SEM, XPS, PL, UV-VIS spectrophotometer and Hall effect measurement. The current-voltage (*I-V*) curve of SnO₂/p-Si heterojunction device was tested and analyzed in detail. SnO₂/p-Si heterojunction shows a prominent visible-lightdriven photoelectrical performance under the low intensity light irradiation. Great photoelectric behavior was also obtained.

Key words: SnO_2 film, chemical bath method, heterojunction, current-voltage (*I-V*) characteristics **PACS**: 78.67. Bf

简易化学水浴法制备 SnO₂ / p-Si 异质结光电性能

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摘要:通过一种简易化学水浴法将 SnO₂ 薄膜沉积在晶硅衬底上以制备 n-SnO₂/p-Si 异质结光电器件,这种自制的化学水浴装置非常便宜和方便.采用 XRD、SEM、XPS、PL、紫外-可见光分光光度计和霍尔效应测试系统 表征了 SnO₂ 薄膜的微结构、光学和电学性能,对 SnO₂/p-Si 异质结的 *I-V* 曲线进行测试并分析,获得明显的光 电转换特性.

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Introduction

Semiconductor heterojunctions have established themselves as the most suitable candidates for optoelectric devices^[1-3]</sup>. Tin oxide (SnO₂) is an important optoelectronic semiconductor with wide direct band gap of 3.6~3.9 eV, which has received enormous attention in recent years. It has attracted considerable attention because of significant applications in the field of optoelectronic devices, such as solar cells, photodiodes, transparent electrodes and photo transistors. There are many merits for SnO_2 , such as abundance in natural resource, low cost and non-toxicity. A variety of techniques have been used to fabricate SnO₂ thin films, such as magnetron sputtering^[4-8], $PLD^{[9-12]}$, ultrasonic spray method^[13-15], sol-gel^[16] and so on. Among these techniques, the chemical bath method is simple, low cost, convenient and fast for large area deposition of polycrystalline SnO₂ films.

In the present study, a polycrystalline SnO_2 film was prepared by a simple chemical bath deposition method on a p-type single crystal Si wafer for making an n-SnO₂/p-Si heterojunction structure photoelectric device. The characterizations of the SnO₂ polycrystalline film were carried out by X-ray diffraction (XRD), scanning electron microscopy (SEM), X-ray photoelectron spectroscopy (XPS), photoluminescence (PL), ultraviolet-visible (UV-Vis) spectrophotometer and Hall effect measurement. The current-voltage (*I-V*) characteristics of the n-SnO₂/p-Si heterojunction were measured by a Agilent 4155C semiconductor parameter analyzer.

1 Experiment

The starting material was a p-type texturized silicon substrate with a boron-doping concentration corresponding to a resistivity of $1\Omega \cdot \text{cm}$. The thickness of the wafer used was 300 μ m and the crystal orientation was (100). The wafer was prepared by a standard cleaning procedure, then it was dipped into a 10% HF solution for one minute to remove the native oxide layer and a texturized. Finally, the wafer was dried in a flow of nitrogen.

SnO₂ thin film was deposited on a Si substrate by a simple chemical bath method to fabricate the n-SnO₂/p-Si heterojunction. The details of the procedure are as follows. Firstly, four tin chloride and hexamethylenetetramine ($(CH_2)_6N_4$, HMT) were dissolved in distilled water by a thermostatic magnetic mixer at 85°C for 10 min. The Molar ratio of four tin chloride and hexamethylenetetramine was 1:1, and the concentration of four tin chloride was vertically placed into the solution and heated at 85°C for 3 hours to grow the polycrystalline SnO₂ film. The sample taken out of the reaction bath was washed with

distilled water and dried with nitrogen. Figure 1 shows the schematic diagram of the polycrystalline SnO_2 film chemical bath deposition system. The thickness of SnO_2 film is about 200 nm.



Fig. 1 Schematic diagram of the polycrystalline SnO_2 film chemical bath deposition system

图 1 多晶 SnO₂ 薄膜简易化学水浴制备系统

Finally, by rod coating, Ag paste was deposited on the back and top sides of SnO_2/p -Si heterojunction. The area of the device is 0.25 cm².

The thickness of SnO_2 film was measured by step profiler (XP-2). The crystallinity, crystal structure, and growth orientations were investigated by X-ray diffraction (XRD). Surface morphology of the SnO_2 film was analyzed with scanning electron microscopy (SEM). The chemical composition was examined by XPS. Photoluminescence (PL) spectra was measured at room temperature. Optical properties of the SnO_2 film were measured by a UV-VIS spectrophotometer in the wavelength rang from 300 nm to 900 nm. The electrical properties of the film were characterized by Hall effect measurement unit (Accent HL5500pc) at room temperature. The currentvoltage (*I-V*) characteristics of the device were measured by an Agilent 4155C semiconductor parameter analyzer.

2 Results and discussion

2.1 Microstructural, optical and electrical properties of SnO_2 films

Crystallinity and structural information of as deposited SnO_2 film were estimated by X-ray diffraction analysis. X-ray diffraction spectra of the SnO_2 film deposited by chemical bath method are presented in Fig. 2 (a). The sample shows a polycrystalline phase. The dominant peak located at 31. 7° is attributed to the SnO_2 (111) diffraction^[17]. The result indicates that the SnO_2 film prapared by chemical bath method has a cubic phase structure with its dominant orientation strongly along the (111) direction perpendicular to the Si substrate surface. This indicates that the SnO_2 film via a simple chemical bath deposition method is polycrystalline in nature and highly textured along the (111) peak was cal-



culated by the Scherrer equation, which is expressed as follows:

Fig. 2 Microstructural and optical properties of SnO₂ films. (a) XRD spectra of the SnO₂ film, (b) XPS survey spectra of the SnO₂ film, (c) Sn 3d narrow scan XPS spectra, (d) O 1s narrow scan XPS spectra, (e) The $(\alpha h\nu)^2$ vs $h\nu$ plot of SnO₂ film, (f) Photoluminescence spectra of the SnO₂ film

图 2 SnO₂ 薄膜的微结构和光学特性. (a) SnO₂ 薄膜 XRD 谱, (b) SnO₂ 薄膜 XPS 全谱图,(c) Sn 3d 的精细 XPS 谱,(d) O 1s 的精细 XPS 谱,(e) SnO₂ 薄膜的($\alpha h\nu$)² vs $h\nu$ 曲线,(f) SnO₂ 薄膜的 PL 谱

$$D = \frac{0.9\lambda}{\beta \cos\theta} \qquad , \quad (1)$$

where λ , θ and β are the X-ray wavelength (1.540 56 Å), the Bragg diffraction angle, and the FWHM of SnO₂-(111) diffraction peak, respectively. The grain size of the SnO₂ film was about 85 nm. Figure 2(a), sinset shows SEM image of the SnO₂ film. The SEM image indicates that the deposited film has a soft surface.

The chemical binding states and surface composition of the SnO₂ film were studied by XPS. Figure. 2 (b) shows the X-ray photoelectron spectroscopy (XPS) survey scan of SnO₂ film on p-type texturized CZ silicon. The sample contains Sn, O, and traces of C. The Sn $3d_5/2$ and Sn $3d_3/2$ peaks shown in Fig. 2 (c) are present at 486. 88 eV and 495. 3 eV, respectively, with a peak splitting of 8. 42 eV, which confirms the presence of Sn⁴⁺. Narrow scan XPS spectra of O 1s state of the sample is shown in Fig. 2 (d). The O 1s peak observed at 530. 73 eV for the SnO₂ film prepared by chemical bath method in the work can be attributed to the chemisorbed oxygen. This result agrees with XRD analysis.

The band gap of SnO_2 film was determined from the

measured transmittance spectra. Figure 2(e) shows the $(\alpha h\nu)^2$ vs $h\nu$ plot of SnO₂ film on glass substrate, which was calculated by a UV-VIS spectrometer. The inset of Fig. 2(e) shows the transmittance curve of SnO₂ film. The average transmittance of the SnO₂ film is about 30% in the visible region of light. The absorption cofficient α were obtained using $\alpha = (1/d) \ln(1/T)$, where *d* is the film thickness and *T* is the transmittance^[18]. The optical band gap, E_g , of the film was calculated using the relation $\alpha h\nu = (h\nu - E_g)^{1/2}$, where $h\nu$ is the photon energy. The optical bandgap is determined by extrapolating the linear part of the $(\alpha h\nu)^2$ curve towards the $h\nu$ axis. In order to evaluate the energy band gap of SnO₂ film, the plot of $(\alpha h\nu)^2$ vs $h\nu$ was shown in Fig. 2(e). The band gap energy (E_g) value of SnO₂ film is 3.88 eV.

Figure 2 (f) shows room-temperature PL spectra of the as-grown polycrystalline SnO_2 film on glass substrate, where the excited wavelength is 325 nm (He-Cd laser). An intensive ultraviolet (UV) emission peak at 390 nm (about 3. 17 eV) and a weak-broad visible emission shoulder at 535 nm (about 2. 32 eV) are observed. The UV emission band could be related to oxygen vacancies of the SnO_2 film^[19], while the weak-broad visible emission basically originates from some defects in the sample.

Electrical properties of the SnO₂ film were determined from the Hall effect measurement. It is an n-type semiconductor. The resistivity is 13.67 $\Omega \cdot \text{cm}$. The electron concentration and mobility are 1.087 × 10¹⁷ atom/ cm³ and 4.201 cm²/V · s, respectively.

 Table 1
 Electrical parameters of SnO₂ film

 表 1
 SnO, 薄膜的电学性能参数

Material	$\rho/(\Omega \cdot cm)$	$n/(\text{atom/cm}^3)$	$\mu/(\mathrm{cm}^2/\mathrm{V}\cdot\mathrm{s})$
SnO ₂	13.67	1.087×10^{17}	4.201

2.2 I-V characteristics

The linear current-voltage (I-V) behaviors between the two Ag electrodes on the surface of the SnO₂ film are shown in Fig. 3. The inset shows the schematic of the test structure. It indicates a good ohmic contact. The distance between the two Ag electrodes on the film is 1 cm.



Fig. 3 $\ I\text{-}V$ characteristics of Ag ohmic contacts to the ${\rm SnO}_2$ film

图 3 Ag 电极与 SnO₂ 薄膜欧姆接触的 I-V 曲线

Figure 4 shows a typical I-V characteristic of the



Fig. 4 *I-V* curve of the SnO₂/p-Si heterojunction in dark 图 4 无光照 SnO₂/p-Si 异质结 *I-V* 曲线

 $\text{SnO}_2/\text{p-Si}$ heterojunction device measured in dark. The inset in Fig. 4 shows the heterojunction device structure. The *I-V* curve of device shows a rectifying behavior. A small leakage current is observed in the reverse bias region, but the forward current is much higher than the reverse current. And the value of $I_{\rm F}/I_{\rm R}$ ($I_{\rm F}$ and $I_{\rm R}$ stand for forward and reverse current, respectively) reaches to 14. 46. In general, the equation of the diode is

$$I = I_0 (e^{\frac{q_r}{nk_B T}} - 1) , \quad (2)$$

where *n* is the ideality factor given by . The value of *n* is found to be 7.4. While the value *n* of AZO/SiO₂/p-Si SIS heterojunction prepared by RF magnetron sputtering is 24. $42^{[20]}$. The results indicate that the SnO₂/p-Si heterojunction has good diode characteristics. The p-n heterojunction is formed at the interface of the p-Si wafer and n- SnO₂ thin film.



Fig. 5 *I-V* characteristic of the SnO₂/p-Si heterojunction in dark and in light (20 W halogen lamp)
图 5 SnO₂/p-Si 异质结无光照及光照(20W 卤钨灯)条
件下 *I-V* 曲线

The photo *I-V* characteristic of the $\text{SnO}_2/\text{p-Si}$ heterojunction device was measured under illumination by a 20 W halogen lamp as shown in Fig. 5. Typical good rectifying and photoelectric behavior are observed for the device. Under the reverse bias, the photocurrent is much

larger than the dark current. For example, when the reverse bias is -5 V, the dark current is only 7.33 $\times 10^{-5}$ A, while the photocurrent reaches to 1.44×10^{-3} A under the 20 W halogen lamp illumination. The photocurrent gain is as high as 19.65, because the visible light can be absorbed in the p-Si substrate to generate many electronhole pairs. Then the electron-hole pairs can be effectively separated by the strong electric field in the interface. Consequently, a large photocurrent is obtained under the reverse bias.

3 Conclusion

The SnO₂ film was successfully deposited on a Si wafer by a simple chemical bath method to fabricate an $n-SnO_2/p-Si$ heterojunction-structured photoelectric device. The self-made chemical bath system is cheap and convenient. The structural, optical and electrical properies of the SnO₂ film were studied by XRD, XPS, UV-VIS spectrophotometer and Hall effect measurement. The *I-V* curves of SnO₂/p-Si heterojunction devices show good rectifying behaviors. This indicates that the p-n heterojunction is formed at the interface of n-type SnO₂ film and p-type Si substrate. A high photocurrent is obtained under a reverse bias. The results indicate that the SnO₂/p-Si heterojunction are promising for low cost visible-light photodiodes for various applications.

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